## JEE MAIN 2020 . TEST-6 SOLUTIONS FUI

STANDARD ANSWER KEY											
Q	1	2	3	4	5	6	7	8	9	10	11
Α	4	1	3	2	4	1	2	3	3	4	2
Q	12	13	14	15	16	17	18	19	20	21	22
Α	2	4	1	4	3	3	4	4	4	2	3
Q	23	24	25	26	27	28	29	30	31	32	33
Α	4	8	3	1	2	2	2	2	4	3	3
Q	34	35	36	37	38	39	40	41	42	43	44
Α	4	3	4	3	4	2	4	3	2	4	3
Q	45	46	47	48	49	50	51	52	53	54	55
Α	2	1	6	3	5	5	1	4	4	1	1
Q	56	57	58	59	60	61	62	63	64	65	66
Α	1	4	1	2	3	2	2	2	4	2	1
Q	67	68	69	70	71	72	73	74	75		
Α	2	1	4	4	3	8	0	1	4		

(1) (4). 
$$(4). \qquad (4). \qquad (1) \qquad (4). \qquad (1) \qquad (1)$$

1L

$$\xrightarrow{\text{weakly}}_{\text{heated}} \underbrace{\bigcirc}_{C-NH_2}^{O} \xrightarrow{\Delta \text{ strongly,}}_{C-NH_2} \xrightarrow{\Delta \text{ strongly,}}_{S73 \text{ K}} \underbrace{\bigcirc}_{O}^{C} \xrightarrow{C}_{C} \\ \underset{O}{}_{NH}$$

(2) (1). 
$$X_2$$
  
 $t = 0$   $\frac{1 \mod 1}{11}$ 

At eqm. 
$$\frac{1-x}{3}$$
  $\frac{2-x}{3}$   $\frac{2x}{3}$ 

 $Y_2 \rightleftharpoons$ 

2 mol

2L

2XY

0

$$\therefore \quad \frac{2x}{3} = 0.6 \text{ mol} / \text{L}. \text{ Therefore, } x = 0.9$$

[X<sub>2</sub>] = 
$$\frac{1}{3} = \frac{1}{3} = 0.3$$
  
[Y<sub>2</sub>] =  $\frac{2 - 0.9}{3} = \frac{2}{3} = 0.3$   
(3) (3). pH = 7 +  $\frac{pK_a - pK_b}{2} = 7 + \frac{0.02}{2} = 7.01$ 

(2).  $C_2H_5OH(\ell) + 3O_2(g)$ (4)  $\rightarrow 2CO_2(g) + 3H_2O(\ell)$ 

In Bomb calorimeter  $\Delta E = -670.48 \text{ kCal mol}^{-1}$  $\Delta H = \Delta E + \Delta n_g RT$  $\Delta H = -670.48 - 1 \times 2 \times 10^{-3} \times 298$ 

(4).  $\mathrm{SO_4^{2-} + BaCl_2 \rightarrow BaSO_4 + 2Cl^-}$ (5) white ppt  $\downarrow$  dil. HCl Not dissolves

The  $\ell$ .p. present on N(II) is involved in resonance and is the part of aromaticity with two pi bonds in ring so to maintain aromaticity the compound does not have tendency to donate this  $\ell$ .p. while  $\ell$ .p. of N (I) is not involved in resonance and is not the part of aromaticity so compound can easily donate this  $\ell$ .p. Protonation is possible on N (I) not on N (II).

(6)

$$\begin{array}{c} CH_{3} & CH_{3} \\ CH_{3} - \overset{}{\underset{C}{C}} - CH_{2} - CH_{2} - OH \xrightarrow{H_{2}SO_{4}} CH_{3} - \overset{}{\underset{C}{C}} = \overset{}{\underset{C}{C}} - CH_{3} \\ CH_{3} & \overset{}{\underset{CH_{3}}{H^{\oplus}}} \\ CH_{3} - \overset{}{\underset{C}{C}} - CH_{2} - \overset{\oplus}{\underset{C}{C}} H_{2} \\ CH_{3} & \overset{}{\underset{C}{C}} H_{3} & \overset{}{\underset{C}{C}} H_{3} \\ CH_{3} & \overset{}{\underset{C}{C}} H_{3} \\ CH_{$$

(8) (3). Given:  
(i) 
$$OC^{1-} + 2H^+$$

(1) 
$$OCI + 2H^{+} + 2e^{-} \rightarrow CI^{-} + H_{2}O^{-}$$
  
;  $E_{1}^{\circ} = 0.94 \text{ V}$ ;  $\Delta G_{1}^{\circ} = -2FE_{1}^{\circ}$ 

(ii) 
$$Cl^{-} \rightarrow \frac{1}{2}Cl_{2} + e^{-}; E_{2}^{\circ} = -1.36 V$$
  
;  $\Delta G_{2}^{\circ} = -1F E_{2}^{\circ}$ 

(iii) OCl<sup>-</sup> + 2H<sup>+</sup> + 2e<sup>-</sup> 
$$\rightarrow \frac{1}{2}$$
Cl<sub>2</sub> + H<sub>2</sub>O;  
; E<sub>3</sub>° = ?;  $\Delta$ G<sub>3</sub>° = -1F E<sub>3</sub>°  
Equation (i) + equation (ii) will gives eq. (iii)  
-2FE<sub>1</sub>° - F E<sub>2</sub>° = -1F E<sub>3</sub>°  
 $\therefore$  E<sub>3</sub>° = 2E<sub>1</sub>° + E<sub>2</sub>°

1

(9) (3). 
$$\operatorname{CHF}_3 \xrightarrow{-H^+} \operatorname{CF}_3^{-1}$$
 (°)  
 $\operatorname{CHCl}_3 \xrightarrow{-H^+} \operatorname{CCl}_3^{-1}$  (°)  
(P\_{\pi}-d\_{\pi}) back bonding more stable.

(10) (4). 
$$SF_6$$
: 'S' atom is strength protested by six  
atoms so hydrolysis is not possible.  
 $NF_3$ : Vacant orbital absent.

(11) (2). ZnO turns yellow on heating as  $Zn^{2+}$  ions move in interstitial sites and electrons also get entrapped in nearby interstitial sites to maintain electrical neutrality. As extra  $Zn^{2+}$  ions are present in interstitial sites thus, it is metal excess defect.

This is known as Liebermann nitroso reaction.

(17) (3). Aldol condensation is not given by compounds which does not have  $\alpha$ -H i.e. methanal

$$\begin{pmatrix} 0\\ \mathbf{H}\\ \mathbf{H}-\mathbf{C}-\mathbf{H} \end{pmatrix}.$$

$$\begin{array}{c} & \bigcirc \\ & \bigcirc \\ & CH_3 - C - H & \longrightarrow & 3\alpha - H \\ & & & \\ & & O \\ & CH_3 - CH_2 - C - H & \longrightarrow & 2\alpha - H \\ & & & O \\ & & & CH_3 - CH_2 - CH_2 - C - H \rightarrow & 2\alpha - H \end{array}$$
 all give aldol condensation

(18) (4). 
$$[Co(py)_3(NH_3)_3]^{3+}$$
 : 2G.I.  
 $[Ni(en)(NH_3)_4]^{2+}$  : No G.I.  
 $[Fe(C_2O)(en)_2]^{2-}$  : 2 optical  
 $[Cr(NO_2)_2(NH_3)_4]^-$  : 3 set of 2 G.I.  
(total 6 isomer)

 $:: NO_2$  can show linkage isomerism

- (19) (4). Larger the value of  $pK_a$ , smaller will be its acidity. Out of the four groups, -COOH, -NO<sub>2</sub> and - CN are e<sup>-</sup> withdrawing which makes benzoic acid more acidic whereas - OCH<sub>3</sub> is e<sup>-</sup> donating which reduces the acidity (makes H<sup>+</sup> less easily available).  $pK_a$  value increases if - OCH<sub>3</sub> is present at paraposition of benzoic acid.
- (20) (4). With trans-but-2-ene, the product of  $Br_2$  addition is optically inactive due to the formation of symmetric meso-compounds.

$$\begin{array}{cccc} H-C-CH_{3} & H & H \\ H & H & H \\ CH_{3}-C-H & H & H \\ Trans-but-2-ene & CH_{3} & CH_{3} \\ \end{array} \xrightarrow{\begin{array}{c} CH_{3} & CH_{3} \\ H & H \\ CH_{3} & CH_{3} \\ \end{array}} Br & Br & H \\ CH_{3} & CH_{3} \\ \end{array} \xrightarrow{\begin{array}{c} CH_{3} \\ H \\ CH_{3} \\ \end{array}} H$$

(21) 2. Volume of hydrogen = 1.12 L

$$\Rightarrow \frac{1.12}{22.4} \times 2g = 0.1 \text{ g}$$
  
0.1 gm H<sub>2</sub> is displaced by  
 $\rightarrow 1.2 \text{ gm of metal}$ 

- $\therefore \quad 1 \text{ gm H}_2 \text{ is displaced by} \\ \rightarrow 12 \text{ gm} = \text{eq. wt. of metal}$
- (22) 3.  $O_2^{-1} \rightarrow 17 e^ \sigma 1s^2 \sigma^* 1s^2 \sigma 2s^2 \sigma^* 2s^2 \sigma 2p_z^2 \pi 2p_x^2$   $= \pi 2p_y^2 \pi^* 2p_x^2 = \pi^* 2p_y^1 \sigma^* 2p_z$ [Electron pair in ABMO = 3]

$$(23) \quad 4. \quad \log\left(\frac{K_2}{K_1}\right)$$

or 
$$\log K_2 - \log K_1 = \frac{Ea}{2.303R} \left( \frac{1}{T_1} - \frac{1}{T_2} \right)$$

1.1 - 2.9 = 
$$\frac{\text{Ea}}{2.303 \times 2}$$
 (1.3 × 10<sup>-3</sup> - 1.5 × 10<sup>-3</sup>)  
(24) 8. T, ° = 353.23 K, W<sub>D</sub> = 1.8 g.

$$W_{A} = 90 \text{ g}, T_{b} = 354.11 \text{ K},$$

$$W_{b} = 2.53 \text{ kg mol}^{-1}$$

$$\Delta T_{b} = T_{b} - T_{b}^{0} = 354.11 - 353.23 = 0.88 \text{ K}$$

$$M_{B} = \frac{W_{B} \times K_{b} \times 1000}{\Delta T_{b} \times W_{A}} = \frac{1.8 \times 2.53 \times 1000}{0.88 \times 90}$$

$$= 57.5 \approx 58 \text{ g mol}^{-1}$$

(25) 3. 
$$\operatorname{Cr}_2\operatorname{O}_7^{2-} + \operatorname{I}^- \to \operatorname{Cr}^{+3} + \operatorname{I}_2$$

- (26) (1). From figure : For time interval t = 0 to t = 1 sec Slope of x-t graph is negative and increasing, so velocity increases in negative direction. For t = 1 to 2 sec. The slope is +ve and decreasing, so velocity is decreasing in +ve direction and become zero at t = 2.
- (27) (2). Maximum heat will be generated when the distance travelled by the block with respect to the belt is maximum this will be the case when the block attains zero velocity after covering a distance  $\ell$  and then come  $\phi$  back.  $a = \mu g$

$$\begin{aligned} \mathbf{v}^2 &= \mathbf{u}^2 - 2\mathbf{a} \; ; \quad \mathbf{0} &= \mathbf{v}_0^2 - 2 \times (\mu g) \ell \\ \mathbf{v}_0 &= \sqrt{2 \mu g \ell} \end{aligned}$$

(28) (2). Spring force does not change instantaneously. Thus for  $m_1$ ;  $a_1 = a_0$ 

> For  $m_2$ :  $F_{S_p} = m_2 a_2$  .....(i) Instantaneously after  $F_2$  is withdrawn Initially  $F_{S_1} = F_2 = m_2 a_2$

From (i) and (ii) 
$$a_2 = \frac{F_2}{m_2} + a_0$$

(29) (2). For equilibrium F = 0 and from F - x graph it is clear at x = 4 F = 0. From V - x graph,

It is clear that 
$$F = \frac{dV}{dx} = 0$$
 at  $x = z$ 

(30) (2). 
$$F_{thrust} - mg = ma$$
  
 $m = 5000 \text{ kg}, a = 20 \text{ m/s}^2$   
 $\Rightarrow F_{thrust} = 150000 \text{ N}$   
 $F_{thrust} = U_{rel} \times \frac{dm}{dt}$   
 $\Rightarrow (-800) \times \frac{dm}{dt} = 150000 \text{ mg}$   
 $\Rightarrow \frac{dm}{dt} \approx -187.5 \text{ kg/s}$ 





in temperature the diameter of both are equal then 4 (1+  $\alpha_s \Delta \theta$ ) = 3.992 (1 + $\alpha_{Brass} \Delta \theta$ ) 4 + 4 × 11 × 10<sup>-6</sup> ×  $\Delta \theta$ = 3.992 + 3.992 × 20 × 10<sup>-6</sup> ×  $\Delta \theta$ 4 + 44 × 10<sup>-6</sup>  $\Delta \theta$ = 3.992 + 79.84 × 10<sup>-6</sup> ×  $\Delta \theta$ 0.008 = 35.84 × 10<sup>-6</sup>  $\Delta \theta$ 

$$\frac{8 \times 10^3}{35.84} = \Delta \theta \quad ; \quad \Delta \theta = \frac{8000}{35.84} = 283$$

So if temperature increased by 223°C then ring will start to slide and this temperature will equal to

 $\theta = 30^{\circ} + \Delta \theta = 30 + 253 = 283^{\circ}C$  $\theta = 283^{\circ}C \approx 280^{\circ}C$ 

(36) (4).





Force on ABCA closed loop zero Force on ADCA closed loop zero Force on extra  $I_1 \& I_2$  $F = (I_1 + I_2) /B = I/B$ 

(40) (4). 
$$\phi = \int \frac{\mu_0 i}{2\pi x} x \, adx = M_x i$$
  
$$M = \frac{\mu_0 a}{2\pi x} (b+c)$$

$$M = \frac{1}{2\pi} \ln \frac{1}{c}$$

(41) (3). 
$$P_{av} = \frac{v^2}{2z_1^2}R = \frac{v^2R}{2z_2^2}$$
 (46)

$$z_1 = \pm z_2 \Longrightarrow \omega^2 L^2 + R^2 = \left(\omega L - \frac{1}{\omega C}\right)^2 + R^2$$
$$-\frac{2L}{C} + \frac{1}{\omega^2 C^2} = 0 \Longrightarrow C = \frac{1}{2\omega^2 L} = 25 \ \mu F$$

(42) (2). 
$$I_1 = I_0 = \frac{1}{4\pi d^2}$$
;  $I_2 = \frac{1}{4\pi (3d)^2}$ 

$$s_1$$
  
 $s_2$   
 $d$   
 $d$ 

$$I = I_1 + I_2 = \frac{P}{4\pi d^2} \left[ 1 + \frac{1}{9} \right] = \frac{10}{9} I_0$$

(43) (4).  $D_1 \Rightarrow$  Reverse biased  $D_2 \Rightarrow$  Forward biased

$$i = \frac{12}{4+2} = 2$$

(44) (3). We know that  $hv = K.E.+\phi$  .....(1)  $\therefore hv_1 = K_1 + \phi$  .....(2) and  $hv_2 = K_2 + \phi$ 

Dividing 
$$\frac{v_2}{v_1} = \frac{K_2 + \phi}{K_1 + \phi} \Rightarrow \frac{2v_1}{v_1} = \frac{K_2 + \phi}{K_1 + \phi}$$
  
 $\Rightarrow 2(K_1 + \phi) = K_2 + \phi \Rightarrow K_2 = \phi + 2K_1$ 

 $\Rightarrow 2 (K_1 + \phi) = K_2 + \phi \Rightarrow K_2 = \phi + 2 K_1$ Hence K.E. will become slightly more than double.

(2). : 
$$T_a = \frac{1}{\lambda}$$
  
 $N = N_0 e^{-\lambda T_a} = N_0 e^{-\frac{1}{T_a} \times T_a} = N_0 e^{-1} = \frac{N_0}{e}$   
 $N = \frac{N_0}{2.718} = 0.37 N_0$ 

(45)

Then number of decayed nuclei  

$$N' = N_0 - N \implies N' = N_0 - 0.37 N_0$$
  
 $N' = 0.63 N_0 \implies 63 \% \text{ of } N_0$ 

or About 2/3 of substance disintegrate.1. Since there is no external horizontal force on

whole system C.M. of whole system need  
move,  

$$\Delta r_{CM} = \frac{m_1 \Delta r_1 + m_2 \Delta r_2}{m_1 + m_2}$$

$$O = \frac{M (x-2) + 3Mx}{4M}$$

$$Mx - 2M + 3Mx = 0; 4x = 2; x = 0.5 m$$
(47) 6.  $w_S - w_E = \frac{2\pi}{T}; w_S - \frac{2\pi}{24} = \frac{2\pi}{8}$ 

$$w_S = 2\pi \left[\frac{1}{2} + \frac{1}{2}\right]; w_S = 2\pi \left[\frac{1+3}{8}\right] = \frac{2\pi}{8}$$

$$w_{S} = 2\pi \left[ \frac{1}{24} + \frac{1}{8} \right] ; w_{S} = 2\pi \left[ \frac{1+3}{24} \right] = \frac{2\pi}{6}$$
  
T<sub>satellite</sub> = 6 hr  
**3.** Let volume of block = V<sub>1</sub>  
Volume of concrete = V<sub>2</sub>

 $\therefore \text{ Displaced volume of water} = (V_1 + V_2)$ Weight of the combination = Buoyant force  $\therefore 0.5 \times V_1 g + 0.25 \times V_2 g = 1 \times (V_1 + V_2) g$ 

$$\therefore \quad \frac{V_1}{V_2} = \frac{3}{1} \quad \therefore \ \frac{m_1}{m_2} = \frac{0.5 \times V_1}{0.25 \times V_2} = 3/5$$

(49) 5. 
$$F - T = 3a$$
;  $T = 2a$   
 $T = 2.5 \times 10^9 \times 4 \times 10^{-8}$ ;  $T = 100 \text{ N}$   
 $T = 2a$ ;  $100 = 2a$ ;  $a = 50 \text{ N}$   
 $F = 5 \times 50$ ;  $F = 250 \text{ N}$ 

(48)

(50) 5. 
$$W = \frac{1}{2} \times 3v_0 \times P_0 + 3v_0 \times P_0$$
$$W = \frac{3}{2} P_0 v_0 + 3v_0 P_0 ; W = \frac{a}{2} P_0 v_0$$
$$\Delta U = nC_v \Delta T = n \left(\frac{3R}{2}\right) (T_f - T_i)$$
$$\Delta U = \frac{3}{2} nR (T_f - T_i) = \frac{3}{2} [P_f V_f - P_i V_i]$$
$$\Delta U = \frac{3}{2} [4P_0 V_0 - 2P_0 V_0] = 3P_0 V_0$$
$$\Delta Q = \Delta U + W = \frac{9}{2} P_0 V_0 + 3P_0 V_0 = \frac{15}{2} P_0 V_0$$
$$P_0 = \frac{10^6}{2} , V_0 = 0.1$$
$$\Delta Q = \frac{15}{2} \times \frac{10^6}{2} \times 0.1 = 375000 J$$
$$\Delta Q = 375 kJ$$
(51) (1). BD = x tanC in  $\Delta PDB$   
and DC = x tanB for  $\Delta PDC$ 
$$\therefore BD + DC = a = x (tanB + tanC)$$
$$A_R = \frac{a}{x} = tanB + tanC \implies result$$

(52) (4). Clearly, equation of required circle, is  

$$(x + 1)^{2} + (y - 1)^{2} + \lambda (x + y) = 0$$

$$\Rightarrow x^{2} + y^{2} + (\lambda + 2)x + (\lambda - 2)y + 2 = 0 \dots (1)$$
As circle (1) intersects the circle  

$$x^{2} + y^{2} + 6x - 4y + 18 = 0 \text{ orthogonally,}$$
so using orthogonality condition, we get

Hence it will never approach  $\infty / -\infty$ 

(2) 
$$f(x) = x^3 + x + 1 \implies f'(x) = 3x^2 + 1$$

- $\Rightarrow$  Injective as well as surjective
- (3)  $f(x) = \sqrt{1+x^2}$  neither injective nor surjective (minimum value = 1) (4)  $f(x) = x^3 + 2x^2 - x + 1$  $\Rightarrow f'(x) = 3x^2 + 4x - 1 \Rightarrow D > 0$

Hence f(x) is surjective but not injective.

(54) (1). 
$$k(x) = \sqrt{1 + \operatorname{sgn} x} = \begin{cases} 0; & x < 0 \\ 1; & x = 0 \\ 2; & x > 0 \end{cases}$$

So, 
$$k(x)$$
 is discontinuous at  $x = 0$ 

(55) (1). 
$$\underset{x \to \infty}{\text{Limit }} \sqrt{x + 1} - \sqrt{x} = 0 \implies \cot^{-1}(0) = \pi/2$$
$$\underset{x \to \infty}{\text{Limit}} \left(\frac{2x + 1}{x - 1}\right)^{x} \rightarrow \infty \implies \sec^{-1}(\infty) = \pi/2$$
$$\therefore l = 1$$

(53)

(56) (1). 
$$u = \int_{0}^{\pi/2} \cos\left(\frac{2\pi}{3}\sin^{2}x\right) dx$$

$$u = \int_{0}^{\pi/2} \cos\left(\frac{2\pi}{3}\cos^{2}x\right) dx \text{ (using King)}$$

$$2u = \int_{0}^{\pi/2} \cos\left(\frac{2\pi}{3}\sin^{2}x\right) + \cos\left(\frac{2\pi}{3}\cdot\cos^{2}x\right) dx$$
(60)
On adding
$$2u = \int_{0}^{\pi/2} 2\cos\frac{\pi}{3}\cdot\cos\left(\frac{\pi}{3}\cos 2x\right) dx$$
(using cos C + cos D)
$$= \frac{1}{2}\int_{0}^{\pi} \cos\left(\frac{\pi}{3}\cos t\right) dt$$

$$[Put 2x = t]$$

$$= \int_{0}^{\pi/2} \cos\left(\frac{\pi}{3}\cos t\right) dt = \int_{0}^{\pi/2} \cos\left(\frac{\pi}{3}\sin t\right) dt$$

$$= v$$
(57) (4). We have f'(x) =  $\sqrt{x} \cos x, x \in \left(0, \frac{5\pi}{2}\right)$ 

$$\stackrel{+}{\longrightarrow} \frac{1}{2} - \frac{1}{2} + \frac{1}{2} +$$

(58) (1). 
$$\mathbf{a} \times (\mathbf{a} \times \mathbf{c}) + \mathbf{b} = 0$$
;  $|\mathbf{a}| = |\mathbf{b}| = 1$ ;  $|\mathbf{c}| = 2$   
 $\mathbf{a} \times \mathbf{d} = -\mathbf{b} \implies (\mathbf{a} \times \mathbf{d})^2 = \mathbf{b}^2 = 1$   
or  $|\mathbf{a}|^2 |\mathbf{d}|^2 - (\mathbf{a} \cdot \mathbf{d})^2 = 1$   
or  $(\mathbf{a} \times \mathbf{c})^2 - 0 = 1 \implies |\mathbf{a}|^2 |\mathbf{c}|^2 - (\mathbf{a} \cdot \mathbf{c})^2 = 1$   
 $\implies 4 - 2\cos^2\theta = 1 \implies \cos^2\theta = \frac{3}{4}; \ \theta = \pi/6$ ]  
Alternative:  $(\mathbf{a} \cdot \mathbf{c})\mathbf{a} - (\mathbf{a} \cdot \mathbf{a})\mathbf{c} = -\mathbf{b}$   
 $(\lambda \mathbf{a} - \mathbf{c})^2 = 1$  or  $\lambda^2 \mathbf{a}^2 + \mathbf{c}^2 - 2\lambda \mathbf{a} \cdot \mathbf{c} = 1$   
(where  $\lambda = \mathbf{a} \cdot \mathbf{c}$ )  
 $\implies \lambda^2 + 4 - 2\lambda^2 = 1$  or  $\lambda^2 = 3$   
 $\mathbf{a}^2 \mathbf{c}^2 \cos^2\theta = 3; \ \cos^2\theta = 3/4; \ \theta = \pi/6$ 

9) (2). The equation of any plane through the intersection of  $P_1$  and  $P_2$  is  $P_1 + \lambda P_2 = 0$  $\Rightarrow (2x-y+z-2)+\lambda(x+2y-z-3)=0$ Also it passes through (3, 2, 1), then  $\lambda = -1$  $\therefore$  Equation of plane is x - 3y + 2z + 1 = 0(3). P(F/F) = 0.9; P(C/F) = 0.1; P(C/C) = 0.80) P(F/C) = 0.2 $P(F) = \frac{3}{10}; P(C) = \frac{7}{10}$ A: Wine tasted was French  $B_1$ : It is a Californian wine;  $P(B_1) = \frac{7}{10}$ С  $B_2$ : It is a French wine;  $P(B_2) = \frac{3}{10}$  $P(A/B_1) = 0.2$ ;  $P(A/B_2) = 0.9$  $P(B_1/A) = \frac{0.7 \times 0.2}{0.7 \times 0.2 + 0.3 \times 0.9}$  $=\frac{0.14}{0.14+0.27}=\frac{14}{41}$ (2).  $x = -\sqrt{-y}$ (61)  $y = -\sqrt{-x} \Rightarrow y^2 = -x$ where x & y both (-) ve  $x = -\sqrt{-y} \implies x^2 = -y$ where x & y both(-) veHence  $A = \frac{16ab}{3}$ , where  $a = b = \frac{1}{4}$ A = 1/3*.*..

(62) (2). Let 
$$P(2t_1, t_1^2)$$
;  $Q(2t_2, t_2^2)$   
Clearly,  $t_1t_2 = -4$  .....(1)  
Also,  $h = \frac{2t_1 + 2t_2}{2}$  .....(2)

3

$$\Rightarrow \frac{3h}{2} = t_1 + t_2$$

and 
$$k = \frac{1 + t_1^2 + t_2^2}{3}$$
 .....(3)

 $3k = 1 + t_1^2 + t_2^2$  $\Rightarrow$ On eliminating we get  $t_1, t_2$  from (1), (2) &(3)

$$, \frac{9h^2}{4} = 3k - 9 \implies x^2 = \frac{4}{3}(y - 3)$$

(63) (2). 
$$O(0,2) \xrightarrow{(4,0)}{F}$$

For the ellipse,  $a^2 = 25$ ,  $b^2 = 9$ 

$$\therefore \quad 9 = 25(1 - e^2) \Rightarrow \ e^2 = \frac{16}{25} \Rightarrow e = 4/5$$

- ... One of the foci is (ae, 0) i.e. (4, 0)
- For the hyperbola ....  $a'e' = 4 \implies 2a' = 4 \implies a' = 2$ and  $b'^2 = 4(e'^2 - 1) = 4 \times 3 = 12$
- Equation of the hyperbola is  $\frac{x^2}{4} \frac{y^2}{12} = 1$ *.*..
- (64) (4). z = (3p - 7q) + i(3q + 7p)for purely imaginary  $3p = 7q \Rightarrow p = 7$ q=3 (for least value) or |z| = |3 + 7i| |p + iq| $\Rightarrow$  |z|<sup>2</sup> = 58 (p<sup>2</sup> + q<sup>2</sup>) = 58 [7<sup>2</sup> + 9] = 58<sup>2</sup> (2). On the set N of natural number,  $R = \{(x, y) :$ (65)  $x, y \in N \text{ and } 2x + y = 41$ Here  $(1, 1) \notin R$  as  $2 \cdot 1 + 1 = 3 \neq 41$ . So, R is not reflexive.  $(1, 39) \in \mathbb{R}$  but  $(39, 1) \notin \mathbb{R}$ . So, R is not symmetric.  $(20, 1), (1, 39) \in R$ .

But  $(20, 39) \notin R$ . So, R is not transitive]

$$(66) \quad (1). \quad \sigma_x^2 = \left\{ \frac{f_1 x_1^2 + f_2 x_2^2}{f_1 + f_2} - \left(\frac{f_1 x_1 + f_2 x_2}{f_1 + f_2}\right)^2 \right\}$$
$$\sigma_x^2 = \left(\frac{1}{f_1 + f_2}\right) \left\{ f_1 x_1^2 + f_2 x_2^2 - \frac{(f_1 x_1 + f_2 x_2)^2}{f_1 + f_2} \right]$$
$$\sigma_x^2 = \frac{1}{(f_1 + f_2)^2} \left( \frac{f_1^2 x_1^2 + f_1 f_2 x_1^2 + f_2 f_1 x_2^2 + f_2^2 x^2}{-f_1^2 x_1^2 - f_2^2 x_2^2 - 2f_1 f_2 x_1 x_2} \right)$$
$$\therefore \quad \sigma_x^2 = \frac{f_1 f_2}{(f_1 + f_2)} (x_1 - x_2)^2$$

(2).  $\sim p$ : Ram does not work hard (67) Use ' $\rightarrow$ ' symbol for then  $(\sim p \rightarrow q)$  means, if Ram does not work hard, then he gets good grade.

(68) (1). Put ln x = t

$$\frac{dx}{x} = dt; \quad \int_{1}^{2010} \left(1 + \frac{1 - t}{t(t - \ell n t)}\right) dt$$

$$2009 + \int_{1}^{2010} \frac{\frac{1}{t} - 1}{t - \ell n t} dt$$

$$2009 - \left[\ell n(t - \ell n t)\right]_{1}^{2010}$$

$$2009 - \left[\ell n (2010 - \ell n 2010)\right]$$
(4)  $y = qx + \beta$  is a tangent

(69) (4). 
$$y = \alpha x + \beta$$
 is a tangent  
Then  $\beta^2 = 4\alpha^2 - 9$   
the locus of P is  $4x^2 - y^2 = 9$  which is hyperbola.  
(70) (4).  $\vec{a} \times \vec{b} + \vec{b} \times \vec{c} = n\vec{a} + d\vec{b} + r\vec{c}$ 

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$$\vec{c} \cdot (\vec{a} \times \vec{b} + \vec{b} \times \vec{c}) = \frac{p}{2} + \frac{q}{2} + r ; p + q + 2r = \pm \sqrt{2} ... (3)$$
Now,  $p = r = -q$ 

$$p = r = \pm \frac{1}{\sqrt{2}}, q = \mp \frac{1}{\sqrt{2}}$$

$$\frac{p^2 + 2q^2 + r^2}{q^2} = 4$$
(71) 3.  $(x - c)^2 = 1 \Rightarrow x - c = \pm 1$ 

$$\Rightarrow x = c + 1, c - 1$$

$$\therefore c - 1 > -2 \text{ and } c + 1 < 4 \Rightarrow -1 < c < 3$$

$$\therefore \text{ Number of integral values of } c \text{ are } 3 \text{ i.e.}, c = 0, 1, 2$$
(72) 8.  $P_n = {n - 2C_3}; P_{n + 1} = {n - 1C_3}$ 

$$Hence {n - 1C_3} - {n - 2C_3} = 15$$

$${n - 2C_3} + {n - 2C_2} - {n - 2C_3} = 15$$
or  ${n - 2C_2} = 15 \Rightarrow n = 8$ 
(73) 0.  $2 \tan^{-1}(1/2) + \tan^{-1}(4/3)$ 

$$= \tan^{-1} \frac{2 \cdot \frac{\pi}{2}}{1 - \frac{1}{4}} + \tan^{-1} \frac{4}{3} = 2 \tan^{-1} \frac{4}{3} > \frac{\pi}{2}$$
  
but  $\operatorname{cosec}^{-1} x \in \left[-\frac{\pi}{2}, 0\right] \cup \left(0, \frac{\pi}{2}\right]$   
 $\Rightarrow \text{ no solution}$ 

1. 
$$A + A^{T} = \mathbf{O} \implies \begin{bmatrix} 2\sin\theta & 0\\ 0 & -2\sin\theta \end{bmatrix} = \mathbf{O}$$
  
 $\implies \sin\theta = 0 \implies \theta = n\pi, n \in I$   
 $\therefore \theta = \pi \in (0, 6)$   
4. We have,  $\frac{dy}{dx} + \left(\frac{-1}{x}\right) y = x (xe^{x} + e^{x} - 1)$   
[Linear differential equation]  
 $\therefore I.F. = e^{-\int \frac{dx}{x}} = e^{-ln x} = \frac{1}{x}$   
Now, general solution is  
 $y\left(\frac{1}{x}\right) = \int (e^{x}(x+1)-1) dx + C$   
 $\implies \frac{y}{x} = x e^{x} - x + C$   
As,  $y (x = 1) = e - 1 \implies \frac{e-1}{1} = e - 1 + C$   
 $\implies C = 0 \qquad \therefore x + \frac{y}{x} = xe^{x}$   
Now,  $2 + \frac{y(2)}{2} = 2e^{2} \implies y(2) = 4e^{2} - 4$ 

(74)

(75)